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## FIRST-YEAR OF MASTER OF SCIENCE CHEMISTRY REVISED SYLLABUS ACCORDING TO CBCS NEP2020

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COURSE TITLE:PHYSICAL & INORGANIC CHEMISTRY PRACTICAL-II  
SEMESTER-II  
W.E.F. 2023-2024

**RECOMMENDED BY THE BOARD OF STUDIES IN CHEMISTRY  
AND  
APPROVED BY THE ACADEMIC COUNCIL**

Devrukh Shikshan Prasarak Mandal's  
Nya. Tatyasaheb Athalye Arts, Ved. S. R. Sapre Commerce, and  
Vid. Dadasaheb Pitre Science College (Autonomous), Devrukh.  
Tal. Sangameshwar, Dist. Ratnagiri-415804, Maharashtra,  
India

Academic Council Item No: 03 dated 08 July 2023

Name of the Implementing Institute	:	Nya. Tatyasaheb Athalye Arts, Ved. S. R. Sapre Commerce, and Vid. Dadasaheb Pitre Science College (Autonomous), Devrukh. Tal. Sangameshwar, Dist. Ratnagiri-415804,
Name of the Parent University	:	University of Mumbai
Name of the Programme	:	Master of Science
Name of the Department	:	Chemistry
Name of the Class	:	First Year
Semester	:	Second
No. of Credits	:	02
Title of the Course	:	Physical and Inorganic Chemistry Practical-II
Course Code	:	S517CHP
Name of the Vertical in adherence to NEP 2020	:	Elective
Eligibility for Admission	:	Chemistry Graduate learner seeking Admission to Post Graduate Programme in adherence to Rules and Regulations of the University of Mumbai and Government of Maharashtra
Passing Marks	:	40%
Mode of Assessment	:	Summative at the end of semester
Level	:	PG
Pattern of Marks Distribution for SEE	:	100 %
Status	:	NEP-CBCS
To be implemented from Academic Year	:	2023-2024
Ordinances /Regulations (if any)		

## Syllabus for First Year of Master of Science in Chemistry

(With effect from the academic year 2023-2024)

### SEMESTER-II

Course Title: Physical & Inorganic Chemistry Practical-II

No. of Credits - 02

Type of Vertical: Elective

COURSE CODE: S517CHP

### Learning Outcomes of Physical Chemistry Practicals Based on BLOOM'S Taxonomy:

After completing the course, the learner will be able to...

Course Learning Outcome No.	Blooms Taxonomy	Course Learning Outcome
CLO-01	Apply	apply phase rule on three component system.
CLO-02	Analyse	analyze different samples of alloys.
CLO-03	Evaluate	determine formula of metal-ammonia complex, CMC and Hammett constant and estimate the amount of different metal cations.
CLO-04	Create	perform standardisation procedures for laboratory instruments and prepare standard solutions of various concentrations.

## Syllabus for First Year of Master of Science in Chemistry

(With effect from the academic year 2023-2024)

### SEMESTER-II

Course Title: Physical & Inorganic Chemistry Practical-II

No. of Credits - 02

Type of Vertical: Elective

COURSE CODE: S517CHP

COURSE CONTENT			
Module No.	Content	Credits	No. of Hours
1	<b>Physical Chemistry Practicals</b>  <b>Non – Instrumental:</b> <ul style="list-style-type: none"><li>To study phase diagram of three component system water – chloroform /toluene -acetic acid.</li><li>To determine the rate constant of decomposition reaction of diacetone alcohol by dilatometric method.</li></ul> <b>Instrumental:</b> <ul style="list-style-type: none"><li>To determine Hammett constant of m- and p- amino benzoic acid/nitro benzoic acid by pH measurement.</li><li>To determine the Michaelis – Menten’s constant value (Km) of the enzyme Beta Amylase spectrophotometrically</li></ul>	1	30
2	<b>Inorganic Chemistry Practicals</b>  <b>Ores and Alloys</b> <ul style="list-style-type: none"><li>Analysis of Tin Solder alloy</li><li>Analysis of Limestone.</li></ul> <b>Instrumentation</b> <ul style="list-style-type: none"><li>Estimation of Fe<sup>+3</sup> solution using Ce (IV) ions Potentiometrically.</li></ul>	1	30
	<b>Total</b>	<b>2</b>	<b>60</b>

### **Access to the Course**

The course is available for all the students admitted for Master of Science.

### **Forms of Assessment**

Vocational Skill Courses, Skill Enhancement Courses and the courses having laboratory sessions shall be assessed at the end of each semester.

### **References:**

1. Practical Physical Chemistry, B. Viswanathan and P.S. Raghavan, Viva Books Private Limited, 2005.
2. Practical Physical Chemistry, A.M. James and F.E. Prichard, 3rd Edn., Longman Group Ltd., 1974.
3. Experimental Physical Chemistry, V.D. Athawale and P. Mathur, New Age International Publishers, 2001
4. Vogel's textbook of quantitative chemical analysis, Sixth Ed. Mendham, Denny, Barnes, Thomas, Pearson education.
5. The Synthesis and Characterization of Inorganic Compounds by William L. Jolly
6. Inorganic Chemistry Practical Under UGC Syllabus for M.Sc. in all India Universities By: Dr Deepak Pant